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## **STUDY ON CONSORTIUM OF BACTERIA IN CHROMIUM LADEN AQUATIC ECOSYSTEM OF INDIA**

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### **ABSTRACT**

#### **Keywords:**

Chromium(VI), chromium  
(III) bacteria, 16S rRNA and  
phylogenetic relationship

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The Chromium is most dangerous heavy metal persistent in the environment. It exists in hexavalent Cr (VI) ion which is mutagenic, carcinogenic and toxic to all life forms. Few microorganisms have developed various mechanisms of adsorption, methylation, oxidation and reduction of chromium and converting it into less toxic trivalent Cr (III) form. These organisms are bacteria, fungi, phytoplankton and zooplankton which form food web based on the physicochemical parameters of that aquatic system. The present study tries to identify all the bacteria occupying aquatic and soil ecological niche laden with chromium. Total of 28 bacteria were isolated and the 16S rRNA were used to study the phylogenetic relationship in these organisms. Bacteria are ranked as most adaptable organism because of its fast transferable plasmid helping in enzymatic transformation of Cr and precipitating metal from aqueous phase. These organisms can be exploited for bioremediation and biosorbent in bioreactors.

## INTRODUCTION

Chromium is one of the most widely used heavy metals in industrial processes (Wang 2000). Chromate is a serious environmental pollutants due to its wide use in industries like corrosion control, pigment manufacturing electroplating, metal finishing, chromate preparation, leather tanning (Ackerley et al., 2004; Thakur et al., 2001; Shaili and Thakur, 2006). It is also used in cooling towers of heavy metal industry, atomic power plants and nuclear weapons production (Barceloux., D. G. 1999).

Chromium exists in nine valence states ranging from -2 to + 6 (Liliana and Eliseo 2008) but in aquatic environment it exists in two stable oxidation states, trivalent chromium and hexavalent chromium. Trivalent chromium Cr (III) is less toxic and mobile.

Hexavalent chromium Cr (VI) is recognized as one of the most dangerous environmental pollutants due to its easy solubility, toxicity and carcinogenic ability to cause mutations and cancer in humans. Cr (VI) typically exists in one of these two forms; chromate ( $\text{CrO}_4^{2-}$ ) or dichromate ( $\text{Cr}_2\text{O}_7^{2-}$ ) (Shen and Wang 1994) and is 100 fold more toxic than trivalent chromium (Meghraj et al., 2003).

The presence of high levels of chromate in the environment also has an inhibitory effect on most microorganisms (Ajmal et al., 1984). However, microorganisms have evolved resistance mechanisms that lead to the selection of resistant variants that can tolerate metal toxicity (Losi and Frankenberger, 1994; Wong and Trevors 1998). There are reports of live microbial systems for the purpose of remediation of contaminated soils and water (Kratochvil et al., 1998). Bioremediation or removal of toxic heavy metal from effluent is generally mediated by diverse group of organisms namely bacteria fungi and planktons (Pandi et al 2007). These microorganisms have the exceptional ability to adapt to and colonize noxious metal polluted environments, which are inhabitable by higher organisms. These microorganisms have developed the capabilities to protect themselves from heavy metal toxicity by various mechanisms such as adsorption, uptake methylation, oxidation and reduction (Megharaj et al 2003). Many questions about the response of complex systems to toxicity are difficult to address by describing different natural environments (Hart 1945). Changes in taxonomic composition and abundance of communities change with response to concentrations of effluents (Koteswari and Ramnibai 2004). The response of taxonomic composition as the end point of microcosm toxicity is reasonably sensitive to many types of toxicants (Plafkin et al., 1989).

The 16s ribosomal RNA gene is a commonly used tool for identifying bacteria for taxonomical studies. Traditional characterization depended upon phenotypic traits like gram positive or gram negative, bacillus or coccus, etc. Taxonomists today consider analysis of an organism's DNA more reliable than classification based solely on phenotypes. We can also identify or classify only the bacteria within a given

environmental samples using 16S rRNA. While there is a homologous gene in eukaryotes, the 18S Rrna gene is relatively short at 1.5 kb, making it faster and cheaper to sequence than many other unique bacterial genes.

In the present study we have tried to pool in the entire bacteria consortium present in the various water bodies in and around the leather and electroplating industries of Tamil Nadu India.

## **MATERIAL AND METHODS**

### **Sample collection:**

In this two year (December 2007 to January 2010) extensive study, water and soil samples were collected and analyzed from chromium laden water bodies, like outlets of leather industry in Ambur District, Common Effluent Treatment Plant (CETP) at Ranipetai and various electroplating industries of Tamil Nadu. The samples were transported to the Environmental Biotechnology Laboratory in black bottles at 4°C.

### **Sample culturing:**

The water samples were cultured on the differential gradient agar of chromium after serial dilution with a non-inoculated differential gradient agar as a control. The agar plates with pH 1 to 12 were also inoculated in order to isolate acidophiles. The bacterial cells were.

### **Sample identification:**

The standard microbial procedures like gram's stain, spore stain were employed to visualize the shape and size of the microorganism. The biochemical tests were performed to identify the bacterial group. The cellular fatty acid chromatography fitted with flame ionization detector, automatic sampler, integrator and computer was used to confirm the results.

### **Bioinformatics Evaluation:**

The partial fragments of 16S RNA were collected from NCBI (online) and were subjected to multiple sequence alignment in BIOEDIT (v\_7.0.5) and parallel verification was performed in CLUSTALW. The tree file generated by the CLUSTALW was visualized using software DENDROSCOPE (v\_2.3) to derive the different patterns of phylograms.

## **RESULTS:**

The extensive study showed 28 organisms overlapping at various sites of chromium rich environments as given in Table 1. Out of these 28 organisms only 10 were found to be gram positive bacteria which were all motile except *Micrococcus*. Except 5 bacteria namely *Agrobacterium*, *Geobacter metallireducens*, *Micrococcus roseu*,

*Micrococcus* and *Streptomyces* all were highly motile. The colony morphology showed that only 5 bacteria colony namely *Bacillus cereus*, *Bacillus subtilis*, *Micrococcus roseus*, *Micrococcus sp.* and *Pseudomonas fluorescens* were flat whereas others had convex elevation. Intriguing colony colors were captivating in these consortia. Among these greenish colonies dominated.

The UTR of these bacteria in Fig 1 and Fig 2 showed many similarities. The specific region of DNA that has proved to be most informative for evolutionary relatedness is 16S Rna, the gene that encodes the RNA component of the smaller subunit of the bacterial ribosome.

## DISCUSSION

The ecological perturbation in an aquatic environment generally produces certain predictable changes in the community structure (Patrick 1971). Species with low tolerance are eliminated, while those species best suited for survival enriched habitats become extensively dominant. Relatively few species were identified which suggests that the early community development was dominated by pioneer species capable of surviving in chromium accumulated environments (Pratt et al., 1993).

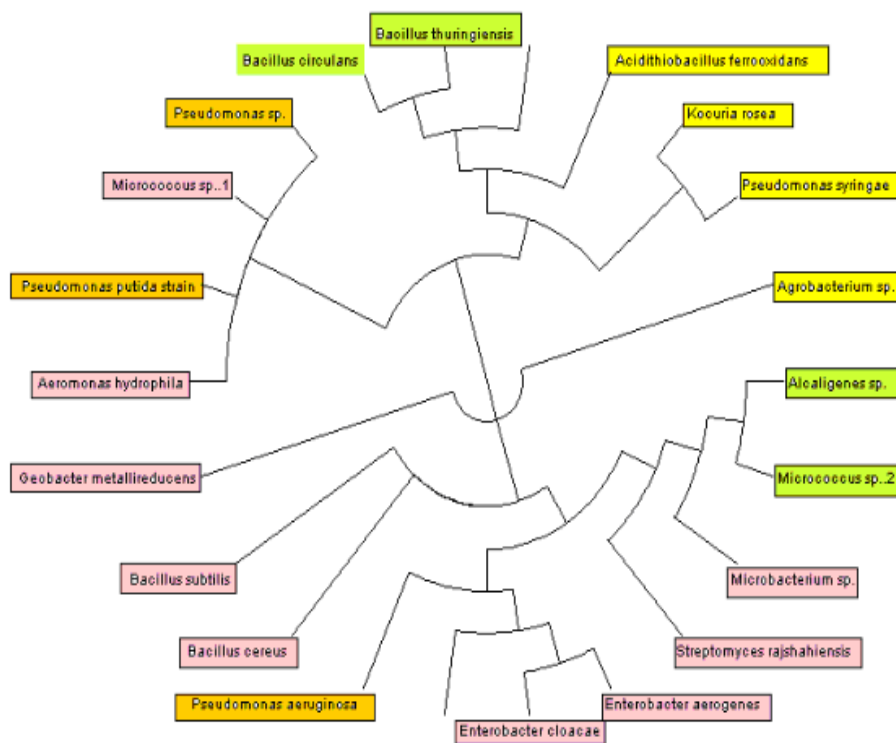
Among the heterotrophic population, bacteria show more tolerance to heavy metals in both aerobic and anaerobic bacteria. The gene encoding resistances to metals tolerance are located at transmissible plasmids. Under environmental conditions of metal stress resistant populations will adapt faster by the spread of R- factors than by mutation and natural selection, thus leading to a very rapid increase in their numbers (Bhattacharjee et al 1988). Bacteria and *Pseudomonas* use Chromium (VI) as terminal electron acceptor during their respiration (Viktoriya et al., 2003). Chromium (VI) is also taken up via sulphate or thiosulphate transporter and oxidized biological molecules resulting in toxicity (Krishna et al., 2004). These microorganisms have developed the capabilities to protect themselves from heavy metal toxicity by various mechanisms such as adsorption, uptake methylation, oxidation and reduction (Megharaj et al 2003). Many questions about the response of complex systems to toxicity are difficult to address by describing different natural environments (Hart 1945 ). Metal ions are converted into insoluble form by specific enzyme reactions and are removed from the aqueous phase (Brierly et al., 1986). The cell surfaces of microorganisms are negatively charged owing to the presence of various anionic structures. This gives microorganisms an ability to bind metal cation. (Chen and Hao, 1998). Intact microbial cell, live or dead and their products can be highly efficient bioaccumulator (Kovacevic et al., 2000).

These organisms discussed in this paper can be used for bioremediation and biosorbent in bioreactors as they continue to do charitable service by adsorbing and transforming chromium from chromium rich ecosystems it into less toxic forms.

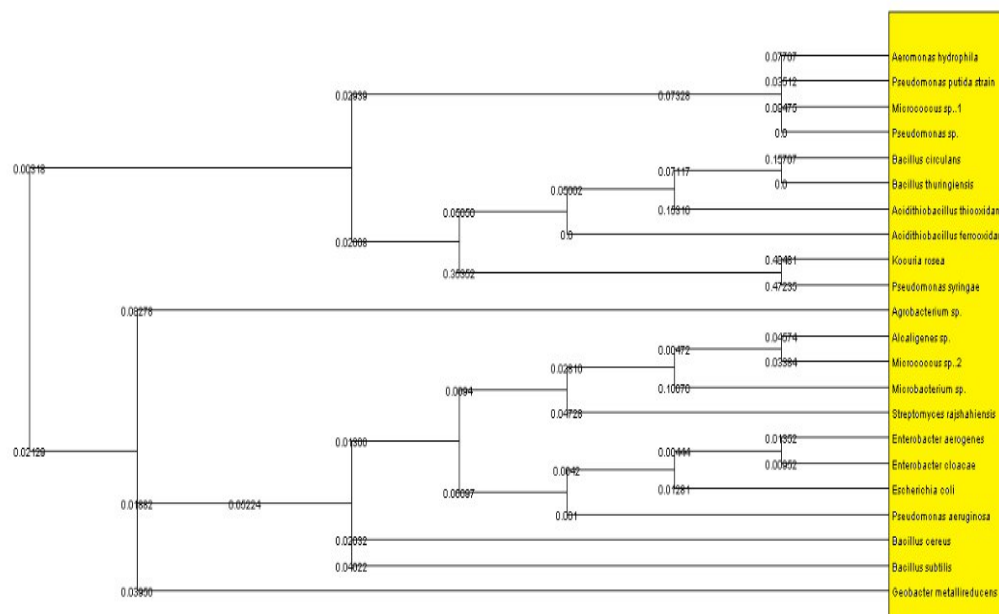
**Table 1: Bacteria consortium in chromium rich aquatic environment**

Name of the organism	Colony color	Colony-form/ Margin/ elevation	Motility Test	Grams Test	Catalase Test
<i>Aeromonasdechromatica</i>	White		Motile		Positive
<i>Aeromonas sp.</i>	White	Circular, Entire,Convex	Motile	-	Positive
<i>Alcaligenes eutrophus</i>	Green	Circular,Entire,Convex	Motile	+	Positive
<i>Agrobacterium</i>	Pink	Circular,Entire,Convex	Non motile	-	Positive
<i>Arthrobacter sp.</i>	White(glossy)	Round,Entire,Convex	Motile	-	Positive
<i>Bacillus cereus</i>	Peacock blue	Irregular,Undulate,Flat	Motile	+	Positive
<i>Bacillus circulans</i>	Pale yellow	Irregular,Wavy,Convex	Motile	+	Positive
<i>Bacillus subtilis</i>	Brown	Circular,Undulate,Flat	Motile	+	Positive
<i>Bacillus thuringiensis</i>	Turquoise	Circular,Entire,Convex	Motile	+	Positive
<i>Desulfovibrio vulgaris</i>	Red	Circular,Entire,Convex	Motile	+	Negative
<i>Desulfovibrio</i>	Red	Circular,Entire,Convex	Motile	-	Negative
<i>desulfuricans</i>	Pale yellow	Silky,Entire,Convex	Motile	-	Positive
<i>Enterobacter cloacae</i>	Pink	Silky,Entire,Convex	Motile	-	Positive
<i>Enterobacter aerogenes</i>	Yellow	Irregular,Entire,Convex	Motile	-	Positive
<i>Escherchia coli</i>	Green		Non motile	-	Negative
<i>Geobacter</i>	Pink	Circular, Undulate, Flat	Non motile	-	Positive
<i>metallireducens</i>	Yellow	Circular,Undulate, Flat	Non motile	+	Positive
<i>Micrococcus roseus</i>	Purple		Motile	+	Positive
<i>Micrococcus sp.</i>	Green	Circular, Entire, Convex	Motile	+	Positive
<i>Microbacterium spp.</i>	Green	Circular, Undulate, Flat	Motile	-	Positive
<i>Pseudomonas sp.</i>	Brown		Motile	-	Positive
<i>Pseudomonas fluorescens</i>	Blue-Green	Irregular,Undulate,	Motile	-	Positive
<i>Pseudomonas ambigua</i>	Blue	Convex	Motile	-	Positive
<i>Pseudomonas aeruginosa</i>	Green	Circular,Entire,Raised	Motile	-	Positive
<i>Pseudomonas putida</i>			Motile	-	Positive
<i>Pseudomonas</i>	Grey		Non motile	-	Positive
<i>chromatophila</i>	Green	Filamentous,Undulate,Co	Motile	-	Positive
<i>Pseudomonas</i>	Blue	nvex	Motile	+	Positive
<i>dechromaticans</i>				-	
<i>Streptomyces</i>					
<i>Thiobacillus thiooxidans</i>					
<i>Thiobacillus ferrooxidans</i>					

**Figure 1. Phylogenetic comparison of UTR of these 28 bacteria**



**Figure 2. Phylogenetic comparison of UTR of these 28 bacteria**



## REFERENCES

1. Ackerley D.F., C.F. Gonzalez, C.H. Park, I.R. Blake, M. Keyhan and A. Matin, 2004. Chromate reducing properties of soluble flavoproteins from *Pseudomonas putida* and *Escherichia coli*, *Appl. Environ. Microbiol.* 70 pp. 873–882.
2. Ajmal, M., Nomani, A.A. and Ahmad A., 1984. Acute toxicity of chrome electroplating wastes to microorganisms: adsorption of chromate and chromium (VI) on a mixture of clay and sand. *Water Air Soil Pollut.* 23 : 119 – 127.
3. Barceloux, D. G. 1999. Chromium. *J. Toxicol. Clin. Toxicol.* 37: 173-194..
4. Bhattacharjee, J.W. , Pathak SP, and Gaur, A., 1988. Antibiotic resistance and metal tolerance of coliform bacteria isolated from Gomti river water Lucknow City. *J. Gen. Appl. Microbiol.* 34. 391 – 399.
5. Brierly, J. A. Goyak, G. M. Brierly, C. L 1986. Consideration for commercial use of natural product for metal recovery. In: Eccles, H. Hunt, S. (Eds.), *Immobilisation of ions by bio-sorption*, Ellis Horwood Chichester, England : 105.
6. Chen, J. M. Hao, O.J. 1998. Microbial chromium (VI) reduction. *Crit. Rev. Environ. Sci. Technol.* 28, 219-225.
7. Hart W. Doudroff P. Greenbank J. 1945. Evaluation of the toxicity of industrial wastes, chemical and other substance to fresh water fishes. *Waste Control Laboratory. Atlantic Refining Co., Philadelphia.*
8. Koteswari Y.N. and Ramanibai R, 2004. Evaluation of toxicity of Tannery Effluent on Plankton community structure: A multispecies microsom study II. *Turk J. Biol.*
9. Kovacevic FZ, L. Sipos and F. Briski, 2000. Biosorption of chromium, copper, nickel and zinc ions onto fungal pellets of *Aspergillus niger* 405 from aqueous solutions, *Food Technol. Biotechnol.* 38 pp. 211–216.
10. Kratochvil, D., Pinentel, P., Volesky, B., 1998. Removal of trivalent chromium by seaweed biosorbent. *Environ. Sci. Technol.* 32, 2693 – 2698.
11. Liliana Morales – Barrera and Eliseo Cristiani – Urbina 2008. Hexavalent chromium removal by *Trichoderma inhamatum* Fungal strain Isolated from tannery effluent. *Water Air Soil Pollut.* 187: 327 – 336.
12. Meghraj M, S. Avudainayagam and R. Naidu, 2003. Toxicity of hexavalent chromium and its reduction by bacteria isolated from soil contaminated with tannery waste, *Current. Microbiol.* 47 , pp. 51–54.
13. Pandi M. Shashirekha V and Swamy M. 2007. Bioabsorption of chromium from retan chrome liquor by cyanobacterial. *Microbial Research.*

14. Patrick R. 1971. The structure and function of fresh water microbial communities. Cairns J Jr. Ed. Research Division Monograph 3. Virginia Polytechnic institute and State University Press Blacksburg Virginia, pp 151-164.
15. Plafkin J.L. Barbour MT, Porter K.D. 1989. Rapid bioassessment protocols for use in streams and rivers: Benthic macroinvertebrates and fishes, U.S. E.P.A. Report – 444/4-89-001.
16. Pratt J R, Bowers N.J , Baiczon JM, 1993. A microcosm using naturally derived microbial communities: Comparative ecotoxicology, In: Environmental Toxicology and Risk Assessment. ASTM STP 1179. Landis WG. Hughes JS. And Lewis MA Eds. Philadelphia: American Society for Testing Materials 178-191.
17. Shaili Srivastava a, Indu Shekhar Thakur b,2006.\*Isolation and process parameter optimization of *Aspergillus* sp.for removal of chromium from tannery effluent , *Bioresource Technology* 97 1167–1173
18. Shen H. and Wang Y.T. 1993. Characterization of enzymatic reduction of hexavalent chromium by *Escherichia coli* ATCC 33456. *Appl. Environ. Microbiol.* 59: 3771-3777.
19. Shen, H. and Wang,Y.T. 1993.Characterisation of Enzymatic reduction of hexavalent chromium by *Escherichia coli* ATCC 33496.*Appl.Environ.Microbiol.*59:3771
20. Thakur.I.S, P. Verma and K.C. Upadhyaya,2001. Involvement of plasmid in degradation of pentachlorophenol by *Pseudomonas* sp. from a chemostat, *Biochem. Biophys. Res. Comm.* 286 , pp. 109–113.
21. Viktoriya V. K., Galyana MD, Rinat RN, Mikhaylo T.B. and Petro I G. 2003. Chromium (VI) reduction in a membrane bioreactor with immobilized *Pseudomonas* cells. *Enzyme and Microbial Technology* 33: 899 -907.
22. Wang, Y. 2000. Microbial reduction of chromate. In D.R. Lovely (Ed.). *Environmental microbe and metal interactions*. Washington. American Society for Microbiology Press. 225 -235.
23. Wong, PTS and Trevors J.T. 1988. Chromium toxicity to algae and bacteria. In Nriagm J.O. and Nieboer, E., Editors, 1998. *Chromium in the natural and Human Environment*, Wiley, New York, 305 – 315.